# Flue Gas Purification Utilizing SOx/NOx Reactions During Compression of CO<sub>2</sub> Derived from Oxyfuel Combustion (Oxy – T – Fired)

## (NETL Cooperative Agreement No. DE-NT0005309)

Department of Energy/National Energy Technology Laboratory (DOE/NETL) 2010 CO2 Capture Technology R&D Meeting, 13-17 September 2010 Pittsburgh, PA, USA

**Kevin Fogash** 

Air Products and Chemicals, Inc.

## Who Is Air Products?

- Global atmospheric, process and specialty gases, performance materials, equipment and services provider
  - Serving industrial, energy, technology and healthcare markets worldwide
- Fortune 500 company
- Known for our innovative culture and operational excellence
- Safety leader in the chemical industry
- Capture techniques
  - Based upon wide experience in ASU, HyCO, combustion applications, cryogenic separations, compression & CO<sub>2</sub> handling
  - Promising proprietary developments point to reductions in cost of CO<sub>2</sub> capture





## Agreement Period of Performance & Cost Share

- Period of Performance:
  - 1 October 2008 30 September 2010

- Air Products
- NETL Cost Share:
- Overall Project Total:

- \$ 251,000 (20%)
- \$ 1,003,995 (80%)
- \$ 1,254,995

- Project Participants: Air Products
- Host Site: Alstom Power Power Plant Laboratories
   Boiler Simulation Facility in Windsor, CT.



## **Technology Fundamentals**

- What is the technology?
- Current status of technology
- Design for the PDU (process development unit)
- Results from PDU campaigns
- Next Steps



## Oxyfuel CO<sub>2</sub> Purification

- Oxyfuel combustion of coal produces a flue gas containing:
  - CO<sub>2</sub> + H<sub>2</sub>O
  - Any inerts from air in leakage or oxygen impurities
  - Oxidation products and impurities from the fuel ( $SO_x$ ,  $NO_x$ , HCl, Hg, etc.)
- Purification requires:
  - Cooling to remove water
  - Compression to 30 bar
    - Integrated SOx/NOx/Hg removal
  - Low Temperature Purification
    - Low purity, bulk inerts removal
    - High purity, Oxygen removal
  - Compression to pipeline pressure



## NOx SO<sub>2</sub> Reactions in the CO<sub>2</sub> Compression System

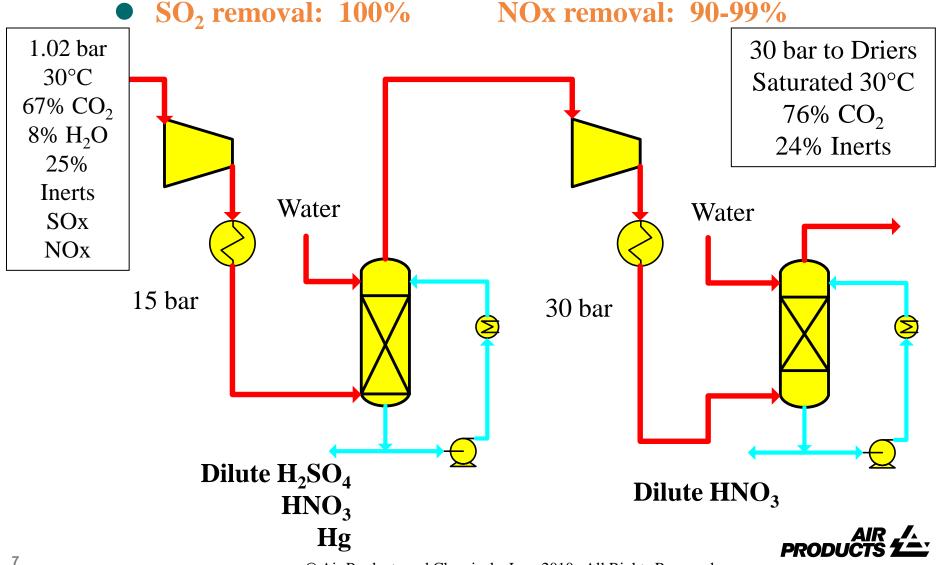
- We realized that SO<sub>2</sub>, NOx and Hg can be removed in the CO<sub>2</sub> compression process, in the presence of water and oxygen.
- SO<sub>2</sub> is converted to Sulfuric Acid, NO<sub>2</sub> converted to Nitric Acid:

$-$ NO + $\frac{1}{2}$ O <sub>2</sub>	=	$NO_2$	(1) Slow
- 2 NO <sub>2</sub>	=	$N_2\bar{O_4}$	(2) Fast
$- 2 NO_{2}^{-} + H_{2}O$	=	$H\overline{NO}_2 + HNO_3$	(3) Slow
-3 HNO2	=	$HNO_{3} + 2 NO + H_{2}O$	(4) Fast
$- NO_2 + \overline{SO}_2$	=	$NO + SO_3$	(5) Fast
$- SO_3 + H_2O$	=	H <sub>2</sub> SO <sub>4</sub>	(6) Fast

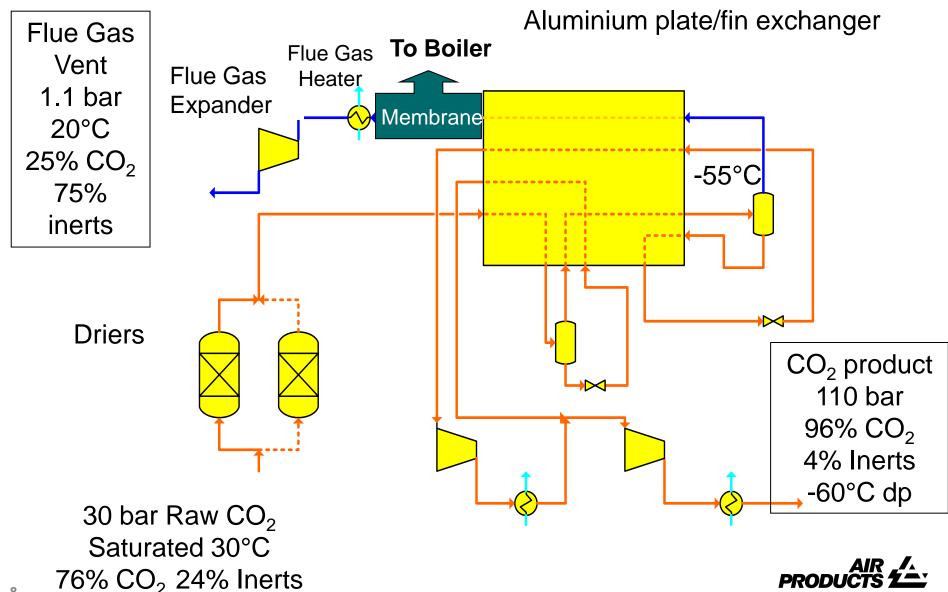
- Rate increases with Pressure to the 3<sup>rd</sup> power
  - only feasible at elevated pressure
- Little Nitric Acid is formed until all the SO<sub>2</sub> is converted
- Pressure, reactor design and residence times, are important.



## Air Products' CO<sub>2</sub> Compression and Purification System: Removal of SO<sub>2</sub>, NO<sub>x</sub> and Hg



### Air Products' System: Inerts removal and compression to 110 bar



## SOx/NOx Removal – Key Features

- Adiabatic compression to 15 bar:
  - No interstage water removal
  - All Water and SOx removed at one place
- NO acts as a catalyst
  - NO is oxidized to NO<sub>2</sub> and then NO<sub>2</sub> oxidizes SO<sub>2</sub> to SO<sub>3</sub>: The Lead Chamber Process
- Hg will also be removed, reacting with the nitric acid that is formed
- FGD and DeNOx systems are not required for emissions or CO<sub>2</sub> purity
  - SOx/NOx removed in compression system
  - Low NOx burners are not required for oxyfuel combustion



**Path to from Lab** 

to Demo



 $160 \, kW_{th}$ oxy-coal rig



 $15\,MW_{th}$ oxy-coal combustion unit

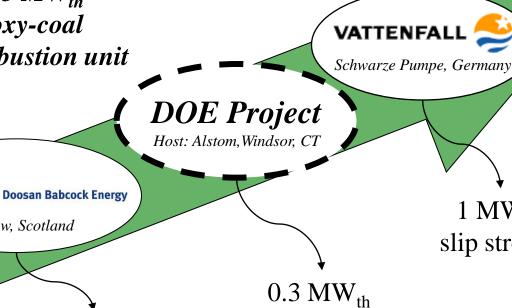
Renfrew, Scotland



30 MW<sub>th</sub> oxy-coal pilot plant



50-250 MW<sub>e</sub> oxy-coal **Demonstration** 



 $0.3~\mathrm{MW_{th}}$ slip stream



 $1 \text{ MW}_{\text{th}}$ 

slip stream



**Imperial College** London London

Batch

DOOSAN

 $6 \text{ kW}_{\text{th}}$ slip stream

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## **Project objectives**

 To purify the CO<sub>2</sub> derived from oxy-coal combustion by utilizing the SOx / NOx reactions that will occur during CO<sub>2</sub> compression

#### Phase 1

Design and Construction of Reactor System for Purification of CO<sub>2</sub> from Oxy-Coal Combustion

- The Phase I objectives include the design, construction, and commissioning of a 15 bar reactor system for removal of SOx /NOx from actual oxy-coal derived, CO<sub>2</sub>-rich flue gas.
- The system will be designed to cool an oxy-coal combustion flue gas slip stream(~0.35 MW<sub>th</sub> flow rate equivalent), compress from 1 bar to 15 bar and react within a 15 bar column the SOx/NOx present in the CO<sub>2</sub> rich flue gas.



## Project objectives (continued)

#### Phase II

Evaluate Robustness of Reactor Performance for Purification CO<sub>2</sub> from Oxy-Coal Combustion

- The Phase II objectives include further evaluations of the reaction process using oxy-coal derived flue gas generated by the host site (Alstom).
  - Evaluate the performance of the reactor based on the reactor effluents for different reactor pressures as well as water recycle rates
  - Characterize the reactor effluents (both liquid and gaseous) to assess any change in reactor performance
- Air Products will develop an engineering model to describe the
   15 bar purification reactor performance.
  - Perform a sensitivity analysis using said model to elucidate those parameters most critical to performance

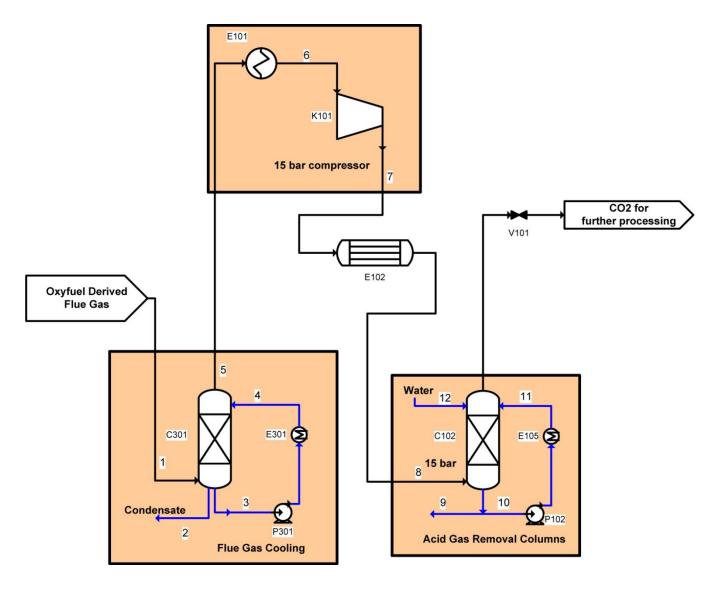


### Milestones / Schedule

- Initiate Construction of Reactor System
  - Planned Date: Complete
- Initiate Testing of Reactor System
  - Planned Date: Complete
- Evaluate Performance of Reactor Based Flue Gas
  - Planned Date: Complete
- Develop Engineering Model and Perform Sensitivity Analysis
  - Planned Date: September 30, 2010



## **Current Process Flow Diagram**



PDU (process development unit)



DOE Project: Air Products' Sour Compression PDU



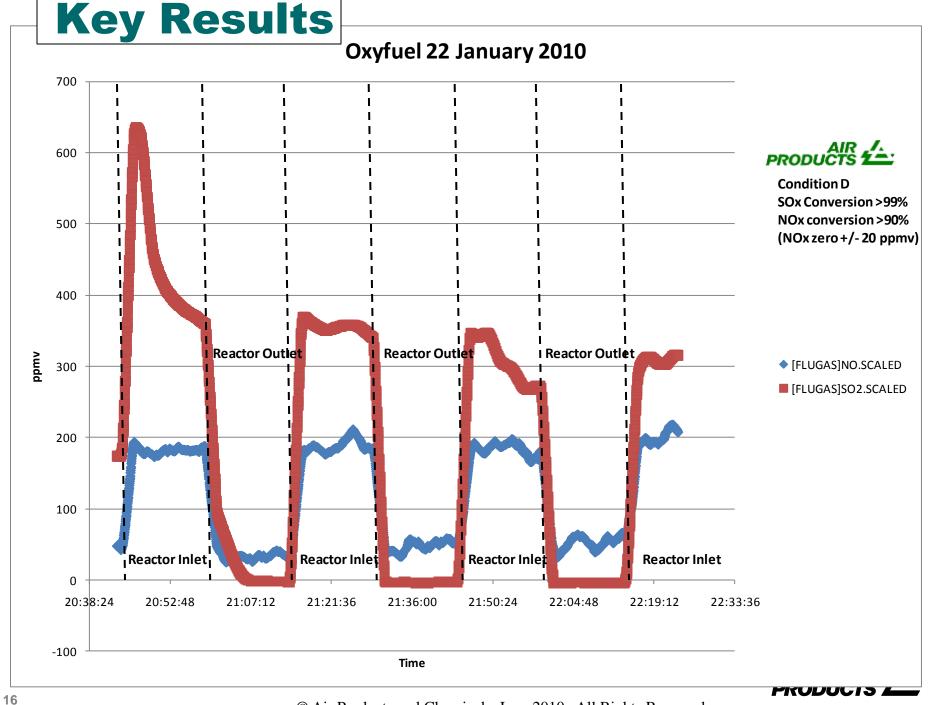


Side View of PDU

- 1st campaign Jan 2010
- 2<sup>nd</sup> campaign April-May 2010

**Acid Reactor (C102)** 



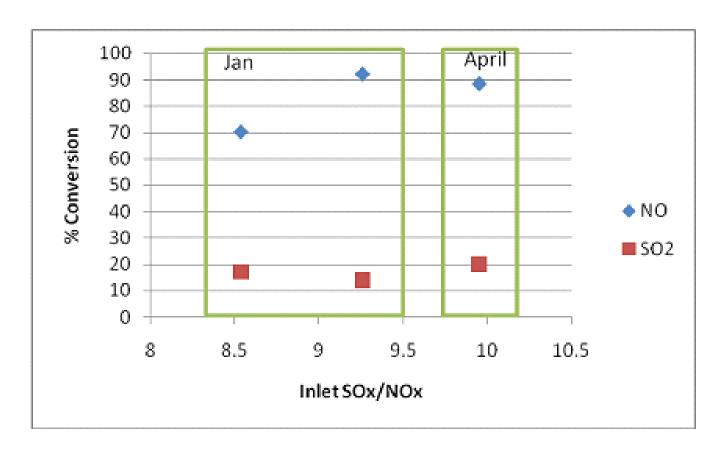


### Results

- For the overall process, total SO<sub>2</sub> removal was 20-100 % (based on gas compositions).
- For the overall process, total NOx removal was 60-90 % (based on gas compositions).
- The effects of variations in the SO<sub>2</sub>/NOx feed ratio, column pressure, gas flowrate and liquid recirculation on the reactor performance were elucidated. Process performance was most sensitive to SO<sub>2</sub>/NOx feed ratio, over the range of parameter values investigated.
- SO<sub>2</sub> was removed from the flue gas through both sulfite and sulfate mechanisms.



## Comparison of NO and SO<sub>2</sub> conversions



- Reproducing test conditions shows consistent results
- Confirms process and measurement reproducibility



## Impact of column operating parameters on SOx and NOx conversion

↑ in Operating Parameter	SO <sub>x</sub> Conversion	NO <sub>x</sub> Conversion
Column pressure	<b>↑</b>	<b>↑</b>
Column gas flow-rate	<b>\</b>	<b>\</b>
Column recirculation liquid flow-rate	1	<b>↑</b>
Column fresh make-up water flow-rate	1	<b>↑</b>
Column inlet SO <sub>x</sub> /NO <sub>x</sub> ratio	<b>\</b>	<u> </u>



## **Advantages**

- FGD and DeNOx systems are not required for emissions or CO<sub>2</sub> purity
  - SOx/NOx removed in compression system
  - Low NOx burners are not required for oxyfuel combustion
- Oxygen can be removed to produce EOR-grade CO<sub>2</sub>
- No penalty if CO<sub>2</sub> is required as a liquid
- Vent stream is clean, at pressure and rich in CO<sub>2</sub> (~25%) and O<sub>2</sub> (~20%)
  - Polymeric membrane unit selective for CO<sub>2</sub> and O<sub>2</sub> in vent stream will recycle CO<sub>2</sub> and O<sub>2</sub> rich permeate stream to boiler.
  - CO<sub>2</sub> Capture increase to >97%
  - ASU size/power reduced ~5%



## Challenges

- Optimization of SOx, NOx, & Hg removal
- Reaction kinetics / equilibrium
- Fouling / impurities effects
- Materials of construction
- Byproduct streams H<sub>2</sub>SO<sub>4</sub>,
   HNO<sub>3</sub>, Hg species,...
- Burners must be demonstrated with flue gas recycle
- Minimization of parasitic power for O<sub>2</sub> supply and CO<sub>2</sub> compression / purification

PDUCPU Pilot Plant

**Boiler OEMs** 

Reference Plants
Design
FEED Studies



Path to from Lab

to Demo



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Batch



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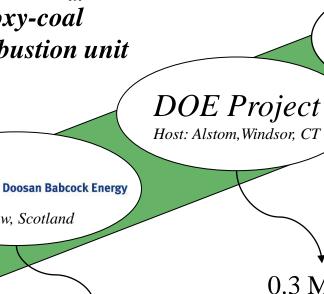
slip stream

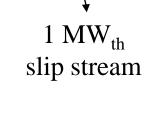


50-250 MW<sub>e</sub> oxy-coal **Demonstration** 

VATTENFALL

Schwarze Pumpe, Germany







**Imperial College** 

 $6 \text{ kW}_{\text{th}}$ slip stream

Renfrew, Scotland



Cylinder fed

bench rig

## **Next Steps**

- Complete final report for PDU / small scale testing of SOx/NOx removal
- Move to pilot scale for CO<sub>2</sub> purification and compression



- Scale up to Pilot: Underway
- Demonstration on stream: 2015
- Commercialization: 2017-2020



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